2)

ION PAIR SOLVATION

FINAL REPORT



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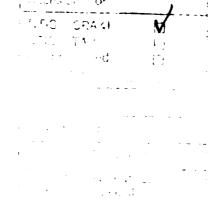
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19 XBSTRACT (Continue on reverse if necessar New tools for the computation	y and identify by block i	number)			e gas phase
and in solution have been developed. Both classical and quantum mechanical methods were					
explored. Monte Carlo techniques for incorporating thermodynamic equilibrium effects of					
surrounding solvent molecule	s have been inco	rporated int	o the comput	er codes.	New
techniques for computing hig					
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The general problem on which our efforts were focused during this grant period dealt with the solvation of ions and ion pairs. We explored the fundamental intermolecular interaction energies among the ion (or ions) and the surrounding solvent molecules. Statistical methods for incorporating the effects of thermodynamic fluctuations in the solvent molecules were also built into our computer simulation capabilities. Finally classical and quantum dynamics tools were used to simulate and study the time evolution of solvent clustered ion pairs which have been excited either thermally or by photochemical means.

The primary result of findings of this grant are detailed in the publications listed below. The names of the people who worked on these projects are also given below. The primary achievement of this effort is the development of state-of-the-art computer simulation techniques for the study of structural, energetic, thermodynamic, and dynamic properties of ions and ion pairs in solution. Earlier progress reports, including the final progress report submitted in August of 1988 provide further details on the specific ions and on pairs which we studied.





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LIST OF MANUSCRIPTS SUBMITTED OR PUBLISHED UNDER ARO SPONSORSHIP DURING THIS REPORTING PERIOD, INCLUDING JOURNAL REFERENCES:

"Rotational Predissociation of Model Atom-Diatom Complexes," R. F. Frey, J. O. Jensen and J. Simons, J. Phys. Chem. 89, 788 (1985).

"Translational and Rotational Symmetries in Integral Derivatives of Arbitrary Order,", J. O. Jensen, A. Banerjee and J. Simons, Chem. Phys. 102, 45-54 (1986).

A Different View of Molecular Electronic Transitions, H. Taylor and J. Simons, J. Phys. Chem. 90, 580.

"Resonance state energies and lifetimes via analytic continuation of stabilization graphs," R. F. Frey and J. Simons, J. Chem. Phys. 84, 4462 (1986).

"Translational and rotational symmetries in third integral derivatives," J. O. Jensen, A. Banerjee and J. Simons, Proc. Ind. Acad. Sci. (Chem. Sci.) 96, Nos 3 & 4, 126 (1986).

"Application of spectral quantization to metastable states of C A'DCN," D. T. Chuljian, J. Ozment and J. Simons, J. Chem. Phys. 85, 5826 (1986).

"Fragmentation Dynamics of Solvent-Clustered Ion Pairs," R. F. Frey and J. Simons, J. Phys. Chem. 91, 2818 (1987).

"Theoretical Study of  $C_2$  and  $C_2^-$ :  $X = \Sigma_g^+$ ,  $X = \Sigma_u^+$ , and  $X = \Sigma_u^+$  and  $X = \Sigma_u^+$  Potentials," J. A. Nichols and J. Simons, J. Chem. Phys. 86, 6972 (1987).

"Monte Carlo Simulations of Small Ion-Pair Hydrate Clusters:

 $NO_2^-:Li^+(H_2O)_n^-$ , A. Banerjee, A. Quigley, R. F. Frey, D. Johnson and Jack Simons, J. Am. Chem. Soc. 109, 1038 (1987).

"Moller-Plesset perturbation theory for van der Waals complexes bound by electron correlation effects: Ground states of the Ar and Mg dimers," G. Chalasinski, D. J. Funk, J. Simons and W. H. Breckenridge, J. Chem. Phys. 87, 3557 (1987).

"Ab initio studies of the structures and energies of the H (H 0) and

 $H^{-}(H_{2}O)_{2}$  complexes," G. Chalasinski, R. A. Kendall and J. Simons, J. Chem. Phys. 87, 2965 (1987).

"Ab initio studies of the structure and energetics of the  $H^-(H_2)$  Complex," G. Chalasinski, R. A. Kendall and J. Simons, J. Phys. Chem. 91, 6151 (1987).

"A unitary multiconfigurational coupled-cluster method: Theory and applications," Mark R. Hoffmann and Jack Simons, J. Chem. Phys. 88, 993 (1988).

SCIENTIFIC PERSONNEL SUPPORTED BY THIS PROJECT AND DEGREES AWARDED DURING THIS REPORTING PERIOD:

Grzegorz Chalasinski Mark Hoffmann Rick A. Kendall Doug O'Neal Hugh Taylor Ramon Hernandez Edward Earl

No advanced degrees were awarded during this reporting period.